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(54) BASICALLY SUBSTITUTED DIBENZOXAZEPINES, DIBENZOTHIAZEPINES AND DIBENZODIAZEPINES

(71) We, Dr. A. Wander S.A., a body corporate organised under the laws of Switzerland of Monbijoustrasse 1.15, 3001 Berne, Switzerland, do hereby declare the invention, for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described in and by the following statement:—

This invention is generally concerned with new heterocyclic compounds. According to the invention there are provided 11-Basically substituted dibenz[b,f]-1,4-oxaze-pines, dibenzo[b,f]-1,4-thiazepines and dibenzo[b,e]-1,4-diazepines of the formula:



(I)

and therapeutically acceptable acid addition salts thereof. In formula I, Z denotes oxygen, sulphur, sulphinyl (—SO—) or imino (—NH—). R₁ represents hydrogen, allyl, alkyl containing not more than 3 carbon atoms, hydroxyalkyl containing not more than 3 carbon atoms, alkoxyalkyl containing not more than 6 carbon atoms or alkanoyloxyalkyl containing not more than 6 carbon atoms: R₂ represents nitro; amino; aminosulphonyl of the formula —SO₂NR₃R₄, wherein R₃ and R₄ are the same or different and represent hydrogen or methyl; alkylsulphinyl of the formula —SOR₅ in which R₅ denotes alkyl with not more than 3 carbon atoms; or alkylsulphonyl of the formula —SO₂R₅ in which R₅ denotes alkyl with not more than 3 carbon atoms.

Compounds of formula I are obtained when a compound of the formula:



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(II)

wherein Z and R2 have the meanings defined above and X denotes a halogen atom or a sulfhydryl, alkoxy, alkylthio, p-mtrobenzylthio or tosyl group is reacted with piperazine or a piperazine derivative, respectively, of the formula:

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 $(\Pi\Pi)$

wherein R_1 has the above-mentioned meaning. Certain of the starting materials of the formula H are obtained by converting lactams of the formula:

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wherein Z and R₂ have the meanings given above, into the thioloctams which may be followed by alkylation, or by reaction of the lactams with a halogeneous agent such as phosphorus oxychloride or phosphorus pentachloride, most suitably in the presence of a catalytic amount of dimethylaniline or dimethylformamide. Lactams of formula IV are themselves obtained by ring closure of compounds of the formula:

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wherein Z and R2 have the above-mentioned meanings and R6 denotes hydrogen or a lower alkyl group containing from 1 to 3 carbon atoms. For products wherein Z represents -O or -S-, lactams of formula IV may also be obtained by ring closure of compounds of the formula:

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wherein Hal stands for halogen, or of isocyanates of the formula:

Lactams of formula IV in which R2 represents amino are most suitably obtained by reduction of the corresponding nitrolactams.

Compounds of formula I may further be obtained by ring closure through intro-

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molecular condensation of acid amides or acid thioamides of the formula:

(VIII),

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wherein Z, R1 and R2 have the above-mentioned meanings and Y represents oxygen or sulphur. A purely thermal condensation rarely succeeds with the acid amides but rather with the thioamides which are, for example, obtained from the acid amides by treatment with phosphorus pentasulphide and need not be isolated before the following 5 condensation. Especially in the case of the acid amides it is desirable to perform the ring closure in the presence of condensing agents, such as for example phosphorus pentachloride, phosphorus oxychloride, phosgene and polyphosphoric acid. It is assumed that the ring closure proceeds by way of intermediate steps such as imidochlorides, amidochlorides, imidophosphates, amidophosphates or salt-like derivatives thereof, 10 which, in general, are not insolatable. The condensation of the thioamides is favoured by the presence of mercury (II) salts or by intermediate formation of imidothioethers. Heating and, if required, the use of a suitable inert solvent, are desirable, and when using phosphorus oxychloride and phosphorus pentachloride, addition of catalytic amounts of dimethylformamide or dimethylaniline. 15

11-Basically substituted dibenz[b,f]-1,4-oxazepines and dibenze[b,f]-1,4-thiazepines (formula I; Z = -0— or -S—) can also be obtained by dehydration of urea derivatives of the formula:

wherein R₂ has the above-mentioned meaning and R₇ means R₁ or denotes a removable group, especially a hydrolytically removable group. The ring closure is preferably carried out by heating in the presence of dehydrating agents such as for example zinc chloride, aluminium chloride, stannic chloride, phosphoric acid and polyphosphoric acid, especially phosphorus oxychloride or phosphorus oxychloride and phosphorus pentoxide, if desired in an inert solvent of suitable boiling point such as for example benzene or toluene. According to the chosen reaction conditions the starting materials of formula IX with a hydrolytically removable group R₇, e.g. carbalkoxy, especially carbethoxy, are cyclized directly to the 11-(1-piperazinyl) compounds by hydrolysis of the removable group. Other removable groups can be split off after ring closure in a way known per se e.g. by hydrogenolysis.

As long as R_2 does not denote amino, the products (I) can also be obtained when amidines of the formula:

wherein Z has the above-mentioned meaning and R'₂ represents R₂ with exclusion of amino, are treated with a reactive ester of an alcohol of the formula:

wherein R₁ has the above-mentioned meaning. The reaction is carried out following or by simultaneous treatment with a basic catalyst or metallization agent such as sodamide, lithium amide, sodium hydride, butyl lithium, phenyl sodium, sodium ethylate or potassium-t-butoxide. Suitable estens are those of inorganic or organic acids, e.g. hydrohalic acid, sulphonic acid or carbonic acid esters. The required amidines X are in turn obtained by treating compounds of formula III with ammonia.

On the other hand, compounds of formula I, wherein R₂ is amino, may be obtained by reduction of the corresponding nitro compounds. The reduction is most suitably carried out by treatment with hydrogen in the presence of a catalyst such as palladium.

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	charcoal or Raney nickel or by treatment with stannous chloride and hydrochlonic acid. Compounds of formula I, wherein Z denotes sulphinyl, are also obtained by oxidation, e.g. with periodates, of the corresponding compounds in which Z represents sul-	
	tion, e.g. with periodates, of the corresponding composition	_
5	Compounds of formula I, wherein R ₂ represents alkylsulphinyl or alkylsulphonyl, respectively, can also be obtained by mild (e.g. with periodates) or strong electronic acid) oxidation of the corresponding alkylthio comhydrogen peroxide or peracepta ellevisulphonyl are also obtainable by strong	5
	hydrogen peroxide or peracent activity by the period of the corresponding alkylsulphonyl are also obtainable by strong pounds. Products wherein R ₂ represents alkylsulphonyl are also obtainable by strong oxidation of the corresponding alkylsulphinyl compounds. If the oxidation is carried out on the dibenzo[b,f]-1,4-thiazepines (Z=—S—) using mild oxidizing agents the out on the dibenzo[b,f]-1,4-thiazepines (Z=—S—) are obtained	10
10	out on the dibenzo[15,1]-1,1-mazepines (25 — SO—) are obtained. corresponding thiazepine sulphoxides Z = —SO—) are obtained.	
	corresponding thiazepune suppostes $Z = SC$ at denotes aminosulphonyl of the form- Finally, compounds of formula I, wherein R_2 denotes aminosulphonyl of the form-	
	Finally, compounds of formula 1, wherein Ky compounds containing the ula —SO ₂ NR ₂ R ₃ , are obtained when the corresponding compounds containing the	
	ula —SO ₂ NR ₂ R ₃ , are obtained when the corresponding some reacted group —SO ₂ X' instead of aminosulphonyl, wherein X' is a halogen atom, are reacted group —SO ₂ X' instead of aminosulphonyl, wherein R ₂ and R ₃ have the above	15
	group — SO_2X' instead of aminosingmonth with all the states of the formula HNR_2R_4 , wherein R_3 and R_4 have the above with ammonia or an amine of the formula HNR_2R_4 , wherein R_3 and R_4 have the above with ammonia or an amine of the formula HNR_2R_4 , wherein R_3 and R_4 have the above with ammonia or an amine of the formula HNR_2R_4 , wherein R_3 and R_4 have the above HR_3R_4 .	15
15	with ammonia or an amine of the formula fittings, which with ammonia or an amine of the formula fittings a sulphochloride group (—SO ₂ Cl) are defined meaning. Starting materials containing a sulphochloride group (—SO ₂ Cl) are	
	defined meaning. Starting materials containing a supercontent of the Meer- obtained by diazotization of the corresponding amino compounds followed by the Meer-	
	win reaction. Compounds of formula I, obtained according to one of the above methods, wherein R ₁ represents hydrogen and wherein R ₂ is not amino can be converted to such com-	20
20	R ₁ represents hydrogen and wherein R ₂ is not aimed by treatment with reactive esters pounds wherein R ₁ does not represent hydrogen, e.g. by treatment with reactive esters pounds wherein R ₁ does not represent hydrogen, e.g. by treatment with reactive esters are	
	pounds wherein R ₁ does not represent hydrogen, e.g. by the mesthonic acid esters are of alcohols of the formula R ₁ —OH. Hydrodalic acid or toluenessulphonic acid esters are of alcohols of the formula R ₁ —OH. Hydrodalic acid or toluenessulphonic acid esters are	
	of alcohols of the formula R_1 —OH. Hydronant active to the introduced by the method of suitable for this purpose. An alkyl group R_1 can also be introduced by the method of suitable for this purpose. An alkyl group R_1 can also be introduced by the method of suitable for this purpose.	
	suitable for this purpose. An alkyl group by the formula and the suitable for this purpose. An alkyl group by the formula aldehydes either with hydrogen reductive alkylation, i.e. by reaction with corresponding aldehydes either with hydrogen reductive alkylation, i.e. by reaction with corresponding agent such as formic acid. The introduc-	25
25	reductive alkylation, i.e. by reaction with corresponding attention and The introduc- in the presence of a catalyst or with a reducing agent such as formic acid. The introduc- in the presence of a catalyst or with a reducing agent such as formic acid. The introduc-	22
43	tion of a hydroxyalkyl group K_1 can also be carried out by	
	ing alkylene oxide.	
	Compounds of formula I in which R ₁ the lotter a hydroxylating agent to obtained R ₂ is not amino can be subsequently treated with an alkanoylating agent to obtained	
	I Possis D represents an alkaliuviux valkyl Aloup. I I I I I I I I I I I I I I I I I I I	30
30	anhydrides are especially suitable as alkanoylating agents.	
	anhydrides are especially suitable as acknowledged anhydrides are especially suitable or can otherwise. The bases obtained in this manner are in most cases crystallizable or can otherwise the bases obtained in this manner are in most cases crystallizable or can otherwise.	
	be distilled in high vacuum without decomposition and supplieric acid, nitric	
	be distilled in high vacuum without decomposition and retail sulphuric acid, nitric acids such as for example hydrochloric acid, hydrobromic acid, sulphuric acid, nitric acid, phosphoric acid, acetic acid, oxalic acid, maleic acid, succinic acid, tartaric acid acid, phosphoric acid, acetic acid, oxalic acid, maleic acid, succinic acid, tartaric acid acid, phosphoric acid, acetic acid, oxalic acid, maleic acid, succinic acid, tartaric acid acid, phosphoric acid, acetic acid, oxalic acid, maleic acid, succinic acid, tartaric acid	35
35	acid, phosphoric acid, aceid to form addition salts which are stable in water, and toluene sulphonic acid to form addition salts which are stable in water.	
	and toluene sulphonic acid to foin another sans which their therapeutically acceptable. The bases obtained in the described manner and their therapeutically acceptable are he used as active substances in phar-	
	acid addition salts are new compounts which can be such substances. They produce a maceuticals or as intermediates for the production of such substances. They produce a maceuticals or as intermediates for the production of such substances. They produce a maceuticals or as intermediates for the production of such substances.	40
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	neuroleptics, sedatives and especially as little which R. denotes nitro show the typical	
	Especially compounds of formina I in which is a suppressible the suppression of the compounds of the suppression of the suppres	
	behaviour pattern for neutrolepites. This maintests paramorphine arragonising effect in sion of locomotor activity, a caraleptic and/or an apomorphine arragonising effect in sion of locomotor activity, The most effective compounds in this respect are the com-	
AE	sion of locomotor activity, a catalepite analysis at a standard in this respect are the commice or rats, respectively. The most effective compounds in this respect are the commice or rats, respectively. The most effective compounds in this respect are the commice or rats, respectively.	45
45	mice or rats, respectively. The most elective combined in the dibenz[b,f] - 1,4 - oxazepine pounds 2 - nitro - 111 - (4 - methyl - 11 - piperazinyl) - dibenz[b,f] - 1,4 - oxazepine pounds 2 - nitro - 111 - (4 - methyl - 11 - piperazinyl) - dibenz[b,f] - 1,4 - oxazepine	
	pounds 2 - mitro - illi - (4 - methyl - ili - piperazinyl) - dibenzo[b,f] - 1,4 - thiazepine, and 2 - nitro - illi - (4 - methyl - ili - piperazinyl) - dibenzo[b,f] - 1,4 - thiazepine, and 2 - nitro - illi - (4 - methyl - illi - piperazinyl) - as well as their therapeutically	
	obtained according to inxample if the 2, respectively,	
	acceptable acid addition same supplies the same with 11 a (1 - sine regintly) residues show simul-	50
50		-
	depressant action is shown phalmiacologically of the compounds 2 - mitro - 11 - (1 - piper-in rats. Especially active in this respect are the compounds 2 - mitro - 11 - (1 - piper-in rats. Especially active in this respect are the compounds 2 - mitro - 11 - (1 - piper-zinyi) - dibenzo-	
	in rats. Especially active in this respect and 2 - mitro - 1.1 - (1 - piperazinyl) - dibenzo- azinyl) - dibenz[b,f] - 1.4 - oxazepine and 2 - mitro - 1.1 - (1 - piperazinyl) - dibenzo- azinyl) - dibenz[b,f] - 1.4 - oxazepine and 2 - mitro - 1.1 - (1 - piperazinyl) - dibenzo-	55
55	The first 1.4 - this zeroine obtained according to azamphe to or 20, 2007	22
	therapeutically acceptable acid addition saws.	
		,
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	by 2 - dimethylaminosuppriority - lil - (4 - methyl - 1 - piperazinyl) - 1,4 - thiazepine, 2 - dimethylaminosuphonyl - ll - (4 - methyl - ll - piperazinyl) -	
	1,4 - thiazepine, 2 - dimetryaminusurphonyl - 11 - (4 - methyl - 11 - piperazinyl) - dibenz[byf] - 1,4 - oxazepine, 2 - methylsulphonyl - 11 - (4 - methyl - 11 - piperazinyl) -	

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dibenz[b,f] - 1,4 - oxazepine and 2 - methylsulphonyl - 111 - (4 - methyl - 1 - piperazinyl) - dibenzo[b,f] - li,4 - thiazepine obtained according to Examples 3, 4, 5 or 26, respectively, and their therapeutically acceptable acid addition salts. The compounds of this invention can be administered in the form of pharmaceuti-5 cal preparations containing, besides the active substance, organic or inorganic solid or 5 liquid carriers suitable for enteral or parenteral administration. The pharmaceutical preparations may be, for example, in the form of tablets, dragees, or solutions for injection, one dosage unit containing from 10 to 25 mg of active substance, depending on its nature, on the route of administration and on the physician's prescription, the effective daily dose amounting to from 5 to 400 mg of active substance. 10 10 The following Examples illustrate the invention: -Example 1 4.9 g of 2 - Nitro - 10/11 - dihydro - 1:1 - oxo - dibenz[b,f] - 1; 4- oxazepine (m.p. 263 °C) and 2 ml of N,N - dimethylaniline are heated in 60 ml of phosphorus oxychloride at reflux for 4 hours. The reaction mixture is then evaporated in vacuo 15 15 to remove the excess phosphorus oxychloride and the residue is decomposed with ice/ water and shaken out immediately with chloroform. The chloroform extracts are washed with dilute hydrochloric acid and water, dried over sodium sulphate and evaporated to dryness in vacuo. The crystalline residue consisting of crude 2 - nitro - ill chloro - dibenz[b,f] - 1,4 - oxazepine is heated at reflux for 6 hours with 6 ml of N -20 20 methylpiperazine in 200 ml of xylene. The organic phase is then shaken out with water and dilute hydrochloric acid. The acid extracts are made alkaline with concentrated soda lye and the base which separates is extracted with chloroform. The chloroform extracts are washed with water, dried over sodium sulphate and evaporated to dryness. 25 The residue is crystallized from chloroform/acetone/petroleum ether and gives 4.7 g. · 25 of 2 - nitro - lll - (4 - methyl - 1 - piperazinyl) - dibenz[b,f] - 1,4 - oxazepine in the form of yellow needles of melting point 192—193°C. EXAMPLE 2 2.0 g of 2 - Nitro - 10,111 - dihydro - 111 - oxo - dibenzo[b,f] - 1,4 - thiazepine (m.p. 270—286°C dec.) and 1 ml of N,N - dimethylaniline are refluxed with 15 ml of 30 30 phosphorus oxychloride for 5 hours after which the reaction mixture is evaporated to dryness in vacuo. The residue is treated with xylene, once again evaporated in vacuo and then refluxed for 16 hours with 15 ml of N - methylpiperazine and 10 ml of dioxane. After evaporating to dryness in vacuo, the residue is distributed between ether and 35 dilute aqueous ammonia solution. The ether phase is washed twice with water and then 35 shaken out with dilute acetic acid. The base is set free from the acid extracts by addition of concentrated ammonia solution and taken up in ether. The ether phase is washed four times with water, dried over sodium sulphate and evaporated. The resinous residue obtained is then dissolved in ether, filtered through aluminium oxide and evaporated. The residue is crystallized from acetone/petroleum ether to give 1.7 g of 2 -40 40 nitro - 11 - (4 - methyl - 1 - piperazinyl) - dibenzo[b,f] - 11,4 - thiazepine in the form of yellow matted needles of melting point 141-142°C. Example 3 4.5 g. of 2 - Dimethylaminosulphonyl - 10,111 - dihydro - 11 - oxodibenzo[b,f] -1.4 - thiazepine (m.p. 283-284°C) and 1.3 ml of N,N-dimethylaniline are refluxed in 45 45 40 ml of phosphorus oxychloride for 4.5 hours. The excess phosphorus oxychloride is then distilled off in vacuo and the residue is dissolved in xylene. The xylene solution is poured onto ice/water, shaken out twice with dilute hydrochloric acid and once with water, dried over sodium sulphate and then concentrated to 100 ml in vacuo. 8 ml of N - methylpiperazine are added and the reaction mixture is refluxed for 4 hours and 50 50 then treated with dilute soda lye and water. The xylene phase is separated and shaken out with dilute hydrochloric acid. The acid extracts are made alkaline with concen-

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trated ammonia solution and the base which separates is extracted with chloroform. After drying over sodium sulphate the chloroform extracts are evaporated in vacuo. The residue is crystallized from acetone/petroleum ether whereby 4.9 g of 2 - dimethyl-

aminosulphonyl - 11 - (4 - methyl - 11 - piperazinyl) - dibenzo[b,f] - 1,4 - thiazepine are obtained in the form of slightly yellow needles of melting point 192—193°C.

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<u> </u>	EXAMPLE 4 1.8 g of 2 - Dimethylaminosulphonyl - 10,11 - dhydro - 11 - oxodibenz[b,f] - 1,4 - oxazepine (m.p. 243—245°C) and 0.6 ml of N.N - dimethylaniline are refluxed in 20 ml of phosphorus oxychloride for 4 hours. The excess phosphorus oxychloride is in 20 ml of phosphorus oxychloride the residue dissolved in xylene and poured onto ice/	5
5	water. The xylene solution is shaken our twice with dilute hydrochloric acid and once water, then dried over sodium sulphate and concentrated to 50 ml in vacuo. 3 ml with water, then dried over sodium sulphate and concentrated for 4 hours and	
10	out with dilute hydrochloric acid. The acid extracts are made alkaline with concentrated out with dilute hydrochloric acid. The acid extracts are made alkaline with concentrated out with dilute hydrochloric acid. The ammonia solution and the base which separates is extracted with chloroform. The ammonia solution and the base which separates is extracted with chloroform. The chloroform extracts are dried over sodium sulphate and evaporated in vacuo. The	10
	residue is crystallized from ether/petroleum ether whereby 1.0 g of supposed in the sulphonyl - 11 - (4 - methyl - 11 - piperazinyl) - dibenz[b,f] - 1,4 - oxazepine of melting point 149—150°C are obtained.	15
15 20	5 g of 2 - Methylsulphonyi - 110,111 - dihydro - 111 - oxo - dibenz[b,f] - 114 - oxa- zepine (m.p. 242—244°C) and 11.8 ml of N.N - dimethylaniline are refluxed in 50 ml of phosphorus oxychloride for 5 hours after which the reaction mixture is evaporated to dryness in vacuo. The residue is treated with xylene, once again evaporated and then dryness in vacuo. The residue is treated with xylene, once again evaporated and then dissolved in xylene and poured onto ice. The aqueous phase is shaken out three times	20
25	water and aqueous somuli chloride a small amount of aluminium oxide. The filtrate is active charcoal and filtered through a small amount of aluminium oxide. The filtrate is active charcoal and filtered through a small amount of aluminium oxide. The filtrate is active charcoal and filtered through a small amount of aluminium oxide. The filtrate is active charcoal and then refluxed with li2 ml of N-methylpiperazine for 6 hours. The reconcentrated and then shaken out twice action mixture is treated with water and concentrated soda lye and shaken out twice	25
30	out with dilute hydrochloric activ. The washed with water and aqueous sodium chloride twice with ether. The ether phase is washed with active charcoal and filtered through solution, dried over sodium sulphate, treated with active charcoal and filtered through a small amount of aluminium oxide. The filtrate is concentrated and a small amount of aluminium oxide. The filtrate is concentrated in treated with petroleum ether. The crystals which precipitate are dissolved in	30
35	acetone and, after concentrating, recrystantiaes of significant point 178—179°C are oxazepine in the form of slightly yellow needles of melting point 178—179°C are	35
	obtained. EXAMPLE 6 3.72 g of 2 - Amino - 2' - (4" - methyl - !!" - piperazinyl - carbonyl) - 4' - nitro - diphenylsulphide (m.p. !184—187°C) and 1 ml of N,N - dimethylamine are refluxed diphenylsulphide (m.p. !184—187°C) and 1 ml of photologic is expected with xylene, once again evaporated and then	40
40	orated to dryness. The residue is treated to dryness. The residue is treated to dryness. The residue is treated and the partitioned between benzene and dilute hydrochloric acid. The base is set free from the partitioned between benzene and dilute hydrochloric acid. The base is set free under ice-cooling, with	45
45	are treated with active chartoan. The chloroform. The chloroform extracts concentrated ammonia solution and taken up in chloroform. The chloroform extracts concentrated ammonia solution and taken up in chloroform. The chloroform extracts are washed with water, dried over sodium sulphate and evaporated. The residue is are washed with water, dried over sodium sulphate and evaporated obtained after dissolved in ether and filtered through aluminium oxide. The residue obtained after evaporation of the solvent is systematically crystallized from acetone/ether/petroleum evaporation of the solvent is systematically crystallized from acetone/ether/petroleum evaporation of the solvent is systematically crystallized from acetone/ether/petroleum	50
50	ether. The first fraction to crystalize - dibenzo[b,] - 1,4 - thiazepine of melting point 11 - (4 - methyl - 1 - piperazinyl) - dibenzo[b,] - 1,4 - thiazepine of melting point 138—141°C is obtained from the more soluble portion. This compound is identical to the product obtained according to Example 2.	
55	EXAMPLE 7 3 g of 2 - (4" - Methyl - 1" - piperazinyl - carbonylamino) - 4' - methylsulphonyl - diphenyloxide (m.p. 1145—1146°C) and a mixture of 2 g of phosphorus pentoxide and 10 ml of phosphorus oxychloride are refluxed for 24 hours. The excess phosphorus oxychloride is then disingled off in vacuo and the residue decomposed with ice/phorus oxychloride is then disingled off in vacuo and the residue decomposed with ice/	55
60	phorus oxychloride is then distilled off in vacuo and the least soda lye and shaken out water. The solution obtained is made alkaline with concentrated soda lye and shaken out with ether. The ether extracts are washed with water and shaken out thoroughly with dilute hydrochloric acid. The acid extracts are made alkaline with concentrated soda lye and shaken out with chloroform. The chloroform extracts are washed with water, dried over sodium sulphate and evaporated to dryness in vacuo. The residue is crystal-	60

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lized from acetone/petroleum ether and gives 1.5 g of 2 - methylsulphonyl - 111 - (4 - methyl - 1 - piperazinyl) - dibenz[b,f] - 1,4 - oxazepine of melting point 1178—179°C identical to the product obtained according to Example 5.

EXAMPLE 8

7.9 g of 2 - Nitro - 111 - amino - dibenz[b,f] - 1,4 - oxazepine (m.p. 238—240°C)

7.9 g of 2 - Nitro - 11 - amino - dibenz[b,f] - 1,4 - oxazepine (m.p. 238—240°C) and potassium-t-butoxide (from 4.0 g of potassium) are stirred together in 40 ml of dimethylsulphoxide for 30 minutes. After addition of 7.5 g of bis - (β - chloroethyl)-methylamine hydrochloride, 11.3 g of potassium iodide and a further 20 ml of dimethylsulphoxide the mixture is stirred for a further 114 hours at 80°C. The reaction mixture is then partitioned between benzene and a large volume of water. The benzene layer is washed with water, then exhaustively extracted with dilute acetic acid. The acetic acid extracts are treated with active charcoal, cooled with ice and made alkaline with concentrated soda lye. The base which is set free is taken up in chloroform. The chloroform solution is washed with water, dried with sodium sulphate and evaporated. The residue is dissolved in benzene and filtered through aluminium oxide. After concentration and dilution with petroleum ether, crystals precipitate which are then recrystallized from chloroform/acetone/petroleum ether to give 4.3 g of 2 - nitro - 111 · (4 - methyl - 1 - piperazinyl) - dibenz[b,f] - 1,4 - oxazepine of melting point 192—194°C which is identical to the product obtained according to Example 11.

Example 9

15.2 g of 2 - Nitro - 111 - (4 - methyl - 11 - piperazinyl - dibenz[b,f] - 11,4 - oxazepine obtained according to Example 11 are hydrogenated with hydrogen in the presence of 11 g of 5% palladium-charcoal in 450 ml of methanol at normal pressure. After take-up of 3.05 11 of hydrogen, the hydrogenation is discontinued and the reaction mixture filtered to remove the catalyst. The filtrate is evaporated in vacuo and the residue taken up in chloroform, filtered through aluminium oxide and concentrated. On addition of petroleum ether, crystals are formed which are separated and recrystallized from chloroform/ether/petroleum ether. 114.1 g of 2 - amino - 111 - (4 - methyl - 11 - piperazinyl) - dibenz[b,f] - 11,4 - oxazepine of melting point 1153—1156°C are obtained.

EXAMPLE 10

10. 10. 5 g of 2 - Nitro - 10. - (4 - methyl - 10 - piperazinyl) - dibenzo[b,f] - 10.4 - thiazepine obtained according to Example 2 are mixed with 24.5 g of stannous chloride and while stirring and cooling with ice, treated dropwise with dilute hydrochloric acid (238 ml of concentrated hydrochloric acid and 100 ml of water). The reaction mixture becomes lighter in colour and a white precipitate is formed. After the addition is complete, the reaction mixture is stirred for a further 20 minutes while cooling, then for 15 minutes at 40°C. The reaction mixture is thereupon made strongly alkaline with

concentrated soda lye and the precipitate taken up in ether. The ether phase is exhaustively shaken out with dilute acetic acid and the base liberated from the acetic acid extracts by addition of concentrated ammonia solution and taken up in ether. The ether phase is washed with water, dried over sodium sulphate and evaporated. The residue is dissolved in ether, filtered through aluminium oxide and evaporated. After crystallization of the residue from ether/petroleum ether, \$10.05 g of 2 - amino - \$110 - (4 - methyl - 1 - piperazinyl) - dibenzo[b.f] - 1,4 - thiazepine are obtained as colourless prisms of melting point \$165-167°C\$.

EXAMPLE 111

A solution of 3.4 g of sodium metaperiodate in 40 ml of water is added in one lot a solution of 5.3 g of 2 - nitro - 111 - (4 - methyl - 1 - piperazinyl) - dibenzo[bf] - 1,4 - thiazepine, obtained according to Example 2, while stirring under ice-cooling. The reaction mixture is then stirred at room temperature for 5 hours and left to stand overnight. After diluting with water and treating with active charcoal, the basic fraction is set free under ice-cooling with concentrated soda lye and taken up in benzene. The benzene solution is washed with water, dried over sodium sulphate and concentrated. The solution is filtered through aluminium oxide and evaporated to dryness. The residue is crystallized from acetone and acetone/petroleum ether to give 4.3 g of 2 - nitro - 11 - (4 - methyl - 1 - piperazinyl) - dibenzo[bf] - 1,4 - thiazepine - 5 - oxide in the form of yellow matted needles of melting point 182—125°C.

EXAMPLE 12

A solution of 3.42 g of sodium metaperiodate in 10 ml of water is given in 3 portions to a stirred solution of 6.24 g of 2 - dimethylaminosulphonyl - 1:1 - (4 - methyl-1 - piperazinyl) - dibenzo[b:f] - 1,4 - thiazepine obtained according to Example 3, in 40 ml of water and 10 ml of glacial acetic acid at 0°C. A precipitate which appears is brought into solution by adding 20 ml of 2 N acetic acid. The reaction mixture is kept

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at room temperature for 24 hours, then made alkaline with concentrated soda lye and shaken out with chloroform. The chloroform extracts are washed with water, dried over sodium sulphate and evaporated to dryness in vacuo. The residue is crystallized from acetone/petroleum ether to give 5.9 g of 2 - dimethylaminosulphonyl - 11 - (4 - methyl - 1 - piperazinyl) - dibenzo[b,f] - 1,4 - thiazepine - 5 - oxide of melting point 208-210°C. EXAMPLE 13 The free base obtained from 6.83 g of 2 - thiomethyl - 111 - (4 - methyl - 1 piperazinyl) - dibenz[b,f] - 11,4 - oxazepine maleate (m.p. 198-201°C) is dissolved in 40 ml of water and 10 ml of glacial acetic acid. This solution is treated dropwise while stirring at 0°C with a solution of 3.42 g of sodium metaperiodate in 10 ml of

water. After the addition is complete, the reaction mixture is left to stand at room temperature for 24 hours, then made alkaline with concentrated soda lye and shaken out with other. The other extracts are washed with water and then exhaustively shaken out with dilute hydrochloric acid. The acid extracts are made alkaline with concentrated soda lye and shaken out with chloroform. The chloroform extracts are washed with water, dried over sodium sulphate and evaporated to dryness in vacuo. The residue is dissolved in acctone and treated with 1.8 g of maleic acid. After concentration and addition of other, crystals precipitate which are recrystallized from methanol/acetone/ ether to give 6.0 g of 2 - methylsulphinyl - Ilil - (4 - methyl - 1 - piperazinyl) - dibenz-[b,f] - 1,4 - oxazepine maleate of melting point 206-207°C.

EXAMPLE 14 A solution of 5.2 g of crude 2 - chlorosulphonyl - 11 - (4 - methyl - 1 - piperazinyl) - dibenz[b,f] - 1,4 - oxazepine, obtained as described below, in 80 ml of chlorotreated dropwise at room temperature with form reaction toluene. The 10% dimethylamine solution in stirred for a further 2 hours at room temperature, then for 1 hour at 40°C and finally evaporated to dryness in vacuo. The residue is taken up in dilute acetic acid, treated with active charcoal and made alkaline with concentrated ammonia solution. The base which separates is taken up in benzene, the benzene solution washed three times with water, dried over sodium sulphate and evaporated. The residue is taken up in benzene and filtered through basic aluminium oxide. The residue obtained after evaporation of the solvent is crystallized from acetone/petroleum ether to give 3.2 g of 2 - dimethylaminosulphonyl - lll - (4 - methyl - 1 - piperazinyl) - dibenz[b,f] - l,4 oxazepine of melting point 148-150°C which is identical to the product obtained according to Example 4.

2 - Chlorosulphonyl - 11 - (4 - methyl - 1 - piperazinyl) - dibenz[b,f] - 1,4 -

oxazepine used as starting material is obtained as follows:

15.4 g of 2 - Amino - 111 - (4 - methyl - 1 - piperazinyl) - dibenz[b,f] - 1,4 - oxazepine (m.p. 153-156°C) are dissolved in 50 ml of glacial acetic acid and 15 ml of 38% hydrochloric acid and diazotized in the usual manner at 0°-5°C with a solution of 3.6 g of sodium nitrite in 6 ml of water. The diazonium solution obtained is added within a few minutes while stirring at 10°C to 40 ml of a 30% solution of sulphur dioxide in glacial acetic acid containing 2 g of cuprous chloride. After the development of nitrogen subsides at room temperature, the reaction mixture is warmed for 15 minutes at 40°C. The reaction mixture is then diluted to 11 with water and treated with active charcoal. While stirring and cooling carefully, the basic fraction is precipitated with concentrated soda lye and taken up in chloroform. The chloroform extracts are washed once with dilute soda lye and once with water, dried over sodium sulphate and evaporated. Crude 2 - chlorosulphonyl - 11 - (4 - methyl - 1 - piperazinyl) - dibenz-[b,f] - 1,4 - oxazepine is obtained as residue.

EXAMPLE 15 5.4 g of 2 - Dimethylaminosulphonyl - Itl - (1 - piperazinyl) - dibenz[b,f] - 1,4 oxazepine obtained according to Example 42 are dissolved in 50 ml of acetone and treated with 1 g of arrhydrous potassium carbonate and 2.24 g of ethyl iodide in 20 ml of acetone and refluxed for 3 hours while stirring. The reaction mixture is then evaporated in vacuo and the residue distributed between dilute soda lye and ether. The ether extracts are washed with water and exhaustively shaken out with dilute aqueous hydrochloric acid. The acid extraots are made alkaline with concentrated soda lye and shaken out with chloroform. The chloroform extracts are washed with water, dried over sodium sulphate and evaporated to dryness in vacuo. The residue is crystallized from acetone/ petroleum ether to give 4.9 g of 2 - dimethylaminosulphonyl - 111 - (4 - ethyl - 1 piperazinyl) - dibenz[b,f] - 1,4 - oxazepine of melting point 160-161°C.

EXAMPLE 16

4.63 g of the same starting material as in Example 15 are dissolved in 80 ml of isopropanol and treated with 1.6 g of anhydrous potassium carbonate, then, while stirring and heating, treated dropwise with 3 g of β - methoxyethyl - p - toluene sulphonic acid ester in 100 ml of isopropanol. After the addition is complete, the mixture is refluxed for 1.5 hours, then evaporated in vacuo. The residue is partitioned between dilute soda lye and ether and the ether extracts exhaustively shaken out with dilute hydrochloric acid. The acid extracts are made alkaline with concentrated soda lye and shaken out with ether. The ether extracts are washed with water, dried over sodium sulphate and evaporated in vacuo. The oily residue is dissolved in warm acetone together with 1.2 g of maleic acid and crystallized by addition of ether. 4.9 g of 2 - dimethylaminosulphonyl - 111 - (4 - β - methoxyethyl - 1 - piperazinyl) - dibenz[b,f] - 1,4 - oxazepine maleate of melting point 124—140°C (decomposition) are obtained.

EXAMPLE 17

4 g of 2 - Dimethylaminosulphonyl - 1il - $(4 - \beta - \text{hydroxyethyl} - 1 - \text{piperazinyl})$ -dibenz[b,f] - 1,4 - oxazepine obtained according to Example 39 are mixed with 30 ml of absolute pyridine and 15 ml of acetic anhydride, the mixture left to stand for one hour at room temperature and then warmed for a short time on the steam bath. The reaction mixture is evaporated in vacuo and the residue diluted with water. The basic fraction is precipitated in the cold with concentrated soda lye and exhaustively extracted with ether. The ether phase is washed with water, dried over sodium sulphate and evaporated. The residue is dissolved in acetone and treated with 1.8 g of maleic acid. After concentration of the solution and addition of ether, crystals precipitate which are recrystallized from acetone/ether to give 3 g of 2 - dimethylaminosulphonyl - 11 - $(4 - \beta - \text{acetoxyethyl} - 1 - \text{piperazinyl})$ - dibenz[b,f] - 1,4 - oxazepine maleate of melting point 155—158°C.

Further products corresponding to formula I given in the following table are obtained by analogous procedures to those given above. In the table Z, R_1 and R_2 have the above defined meaning. In the column on the right hand side at means acctone, e= ether, ch= chloroform, me= methanol and pe= petroleum ether.

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Melting Point	base: 190—192°C (from ac/pc)	maleate: 155—156°C (from me/ac)	base: 153—155°C (from ac/pe)	base: 130—134°C (from ac/pe)	base: 186—188°C (from ch/e)	base: 193—195°C (from ac/pe)	base: 189—191°C (from ac/e/pe)	base: 181—183°C (from ethyl acetate/e)	base: 219—223°C (from ac/pe)	base: 180—183°C (from ch/pe)	base: 169—170°C (from ch/pe)	maleate: 180—185°C. (from me/ac/e)	base: 196—197°C (from ch/pe)
Ra	-NO ₂	-NO ₂	-NO ₂	-NO ₂	-SO ₂ N(CH ₈) ₂	-SO ₈ N(CH ₈) ₈	-SO ₂ CH ₃	-NH.	-SO ₂ CH ₈	—SO,CH,	-SO ₂ C ₂ H ₅	—SO ₂ C ₂ H ₆	-SO ₂ C ₂ H ₅
R1	н	—СН ₂ —СН ₃ —ОН	н	-CH2-CH2-OH	н	-CH _s	Ħ	H	-CH _s	н	-CH ₃	Н	-CH ₃
Z	>	>	S	S	S	HN	>	>°	S	S	S	S	0
Example	18	19	20	21	22	23	24	25	26	27	28	53	30

Melting Point	base: 110—112°C (from ac/pe)	base: 147—150°C (from ac/pe)	dihydrobromide: 225—230°C (dec.; from me/ethyl acetate)	dihydrobromide: 233—248°C from me/ethyl acetate)	base: 218—222°C (from ac/pe)	base: 168—170°C (from(ac/pe)	base: 130—133°C (from ac/pe)	base: 102—104°C (from e/pe)	base: 164—166°C (from ac/e/pe)	base: 174—176°C (from ac/pe)	base: 150—151°C (from ac/pe)	base: 181—182°C (from ac/pe)
R ₂	-NO ₂	—SO ₂ N(CH ₃) ₂	—SO ₂ CH ₃	-SO ₂ CH ₃	—SO ₂ NHCH ₃	-SO ₂ NHCH ₃	-SO ₂ C ₂ H ₅	-NO ₂	—SO ₂ N(CH ₃) ₂	-NO ₃	—SO ₂ N(CH ₃) ₂	—SO ₂ N(CH ₃) ₂
R ₁	—CH ₃	Н	-CH ₃	Н	Н	—СН _з	Н	—CH ₂ —CH ₂ —OCH ₃	-CH ₂ -CH ₂ -OH	н	—CH ₂ —CH=CH ₂	H
Z	HN	HN	HN	HN	S	S	>	>	>0	os	>°	\o_\
Example	31	32	33	34	35	36	37	38	39	40	41	42

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5	Production of tablets For the manufacture of tablets, the products of this invention can be mixed with lactose and granulated with water, 0.5% sodium alginate or 1% gelatine solution. The dried granulate is compressed into tablets in the presence of about 5% of takeum, 5% of corn starch and 0.1% of magnesium stearate. In this way, there are obtained, e.g. tablets of the following composition:	5
	A) 2-Nitro-11-(4-methyl-1-piperazinyl)-	
	dibenz[bf]-1,4-oxazepine 25 mg	
10	Lactose 105 mg	10
10	Corn starch 7.5 mg	
	1 arctin	
	wagicathi stanto	
15	These 1155 mg tablets, which are provided with a crack-line, can be administered orally in a dosage of one half to two tablets two to four times per day in the treatment of subjects suffering from any form of schizophrenia, any form of mania, severe psychotic and non-psychotic states of excitement, chorea, athetosis, and extrapyramidal movement disorders.	15
	B)	
20	2-Nitro-1/1-(1-piperazinyl)-dibenz[0,1]-	20
20	1.4-oxazepine	
	Lactose 120 mg	
	Cour States	
25	Takum 7.5 mg Magnesium stearate 0.115 mg	25
	These 155 mg tablets, which are provided with a crack-line, can be administered orally in the dosage of one half to two tablets two to five times, in some cases up to 5 times 4 tablets per day in the treatment of subjects suffering from states of mental depression and especially agitated forms of depression.	
30	C)	30
	2-Dimethylaminosulphonyl-11-(4-methyl-1- piperazinyl)-dibenz[b,f]-1,4-oxazepine ii0 mg Lactose 70 mg Corn starch 5 mg	
35	Talcom 5 mg	35
	Magnesium stearate 0,11 mg	
40	These 90 mg tablets, which are provided with a crack-line, can be administered orally in a dosage of one half to two tablets one to three times per day in the treatment of subjects suffering from nausea and vomiting following operations or ray treatment or due to stomach or metabolism disorders, intoxications, drug incompatibility, pressure on the brain or pregnancy. These tablets may also be used prophylaotically against post operative vomiting.	40
45	WHAT WE CLAIM IS:— 11. Ill-Basically substituted dibenz[b,f]-1,4-oxazepines, dibenzo[b,f]-1,4-thiazepines and dibenzo[b,e]-1,4-diazepines of the formula:	45
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wherein Z represents an oxygen or sulphur atom or a sulphinyl or imino group; R₁ represents a hydrogen atom, an allyl radical, an alkyl radical containing not more than 3 carbon atoms, a hydroxyalkyl radical containing not more than 3 carbon atoms, an

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alkoxyalkyl radical containing not more than 6 carbon atoms or an alkanoyloxyalkyl radical containing not more than 6 carbon atoms; and \mathbb{R}_2 is a nitro or an amino group, an aminosulphonyl group of the formula — $SO_2NR_3R_4$ wherein \mathbb{R}_3 and \mathbb{R}_4 , which may be the same or different, are hydrogen atoms or methyl groups, or \mathbb{R}_2 represents an alkyl-sulphinyl group of the formula — SOR_5 wherein \mathbb{R}_5 denotes an alkyl radical with not more than 3 carbon atoms, or an alkylsulphonyl group of the formula — SO_2R_5 wherein \mathbb{R}_5 denotes an alkyl radical with not more than 3 carbon atoms, and therapeutically acceptable acid addition salts thereof.

2. 2 - Nitro - 11 - (4 - methyl - 1 - piperazinyl) - dibenz[b,f] - 1,4 - oxazepine and its therapeutically acceptable acid addition salts.

3. 2 - Nitro - 1.1 - (4 - methyl - 1 - piperazinyl) - dibenzo[b,] - 1,4 - thiazepine and its therapeutically acceptable acid addition salts.

4. 12 - Nitro - 11 - (piperazinyl) - dibenz[b,f] - 1,4 - oxazepine and its therapeutically acceptable acid addition salts.

5. 2 - Nitro - 1:1 - (1 - piperazinyl - dibenzo[b,f] - 1,4 - thiazepine and its therapeutically acceptable acid addition salts.

6. 2 - Dimethylaminosulphonyl - 11 - (4 - methyl - 1 - piperazinyl) - dibenzo-[b,f] - 1,4 - thiazepine and its therapeutically acceptable acid addition salts.

7. 2 - Dimethylaminosulphonyl - 1|1 - (4 - methyl - 1 - piperazinyl) - dibenz-

[b,f] - 1,4 - oxazepine and its therapeutically acceptable acid addition salts.

8. 2 - Methylsulphonyl - ill1 - (4 - methyl - 1 - piperazinyl) - dibenz[b,f] - 1,4 -

oxazepine and its therapeutically acceptable acid addition salts.

9. 2 - Methyls: dphonyl - lll - (4 - methyl - 1 - piperazinyl) - dibenzo[b,f] - 1, 4- thiazepine and its therapeutically acceptable acid addition salts.

10. A process for the preparation of 11-basically substituted dibenz[b,f] - 1,4 - oxazepines, dibenzo[b,f] - 1,4 - thiazepines and dibenzo[b,e] - 1,4 - diazepines of the formula given in claim 1 in which a compound of the formula:

wherein Z and R₂ have the meanings given in claim 1 and X denotes a halogen atom or a sulphydryl, alkoxy, alkylthio, f-nitrobenzylthio or tosyl group, is reacted with piperazine or a piperazine derivative of the formula:

wherein R1 has the meaning given in claim 1.

11. A process for the preparation of lil-basically substituted dibenz[b,f] - 1,4 - oxazepines, dibenzo[b,f] - 1,4 - thiazepines and dibenzo[b,a] - 1,4 - diazepines of the formula given in claim 1 in which an acid amide or thioamide of the formula:

wherein Z, R₁ and R₂ have the meanings given in claim 1 and Y represents an oxygen or sulphur atom, is subjected to intramolecular condensation.

12. A process for the preparation of !li!-basically substituted dibenz[b,f] - 1,4 - oxazepines and dibenzo[b,f] - 1,4 - thiazepines of the formula given in claim !! wherein Z is an oxygen or sulphur atom, in which a urea derivative of the formula:

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wherein R2 has the meaning given in claim 1 and R7 has the same meaning as R1 or denotes a readily removable group, is subjected to dehydration, if necessary with subsequent splitting off of the removable group.

13. A process for the preparation of 11-basically substituted dibenz[b,f] - 1,4 oxazepines, dibenz[b,f] - 1,4 - thiazepines and dibenz[b,e] - 1,4 - diazepines of the formula given in claim 1 wherein R2 is a nitro group or an aminosulphonyl, alkylsulphinyl or alkykulphonyl group of formulae -SO₂NR₃R₄, -SOR₆ or -SO₂R₅ respectively in which Rs, R, and R, have the meanings given in claim il, in which an amidine of the formula:

wherein Z has the meaning given in claim 11 and IR'2 represents a nitro group or an aminosulphonyl, alkylsulphinyl or alkylsulphonyl group of the formulae -SO2NR3R4 SOR, or SO, R, respectively wherein R, R, and R, have the meanings given in claim 1, is reacted with a reactive ester of an alcohol of the formula:

$$\begin{array}{c} OH-CH_2-CH_2 \\ > N-R_1 \\ OH-CH_2-CH_2 \end{array}$$

wherein R, has the meaning given in claim il.

14. A process as claimed in any one of claims 10 to 13 in which, when R2 is a nitro group, the product is subsequently reduced to give a compound in which R2 represents an amino group.

15. A process as claimed in any one of claims 10 to 13 in which, in the preparation of a compound of the formula claimed in claim 1 in which R2 represents an alkylsulphinyl or alkylsulphonyl group a reactant containing a precursor of the group R2 is used, which precursor is a thioalkyl group of formula SR, R, having the meaning given

in claim 1 which precursor is subsequently oxidised to the alkylsulphinyl or alkylsulphonyl group, or an alkylsulphinyl group which is subsequently oxidised to an alkylsulphonyl group.

16. A process as claimed in any one of claims 10 to 13 in which, in the preparation of a compound of the formula claimed in claim 1 in which R2 represents an aminosulphonyl group, a reactant containing a precursor of the group R_2 is used, which precursor is a group of formula $-SO_2X'$ wherein X' denotes a halogen atom and is subsequently reacted with ammonia or an amine of the formula HNR₈R₄ wherein R₃ and R, have the meanings given in claim 1.

117. A process as claimed in any one of claims 10 to 16 in which, in the preparation of a compound of the formula claimed in claim 1 in which Z represents a sulphinyl group, a compound of said formula in which Z represents a sulphur atom is first prepared and is subsequently oxidised.

18. A process as claimed in any one of claims 10 to 17 in which, in the preparation of a compound of the formula claimed in claim 1, a compound of said formula in which R1 represents a hydrogen atom and R2 is not amino is first obtained, and the hydrogen arom is subsequently replaced by an alkyl group, an alkyl radical containing not more than 3 carbon atoms, a hydroxyalkyl radical not containing more than 3 carbon atoms, an alkoxyalkyl radical containing not more than 6 carbon atoms or an alkanoyloxy alkyl radical containing not more than 6 carbon atoms.

19. A process as claimed in claim 18 in which the hydroxyalkyl group R1 is sub-

sequently alkanoylated.

	20. A process as claimed in any one of claims 10 to 19 in which the product is isolated as an acid addition salt.	
	21. A process as claimed in any one of claims 10 to 19 in which the product is isolated as the free base.	
5	22. A process for the preparation of compounds of the formula given in claim 1 substantially as herein described with reference to Example T.	. 5
	23. A process for the preparation of compounds of the formula given in claim 1 substantially as herein described with reference to Examples 2, 18 and 19.	
10	24. A process for the preparation of compounds of the formula given in claim 1 substantially as herein described with reference to Examples 3 and 4	
	23. A process for the preparation of compounds of the formula given in claim 1 substantially as herein described with reference to Examples 5, 9, 40, and 20 to 24	10
	20. A process for the preparation of compounds of the formula given in claim 1 substantially as herein described with reference to Examples 25 to 34	
15	27. A process for the preparation of compounds of the formula given in claim 1 substantially as herein described with reference to Examples 6 to 8, 11 to 17, and	15
	28. Dibenzo oxazepines, dibenzo thiazepines and dibenzo diazepines an eleimod	
20	29. Therapeutic compositions comprising a compound as chained in along it is	22
	30. Compositions as claimed in claim 29 in which one dosage unit contains from 10	20
<u></u>	31. Therapeutic compositions as claimed in claim 29 substantially as herein dea	
25		25
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	52/54 High Holborn, London W.C.1. Agents for the Applicants.	

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